(malachite), $2\text{Co}(\text{CO}_3) \cdot 3\text{Co}(\text{OH})_2 \cdot \text{H}_2\text{O}$, $\text{Co}_{0.69}\text{Fe}_{0.34}(\text{CO}_3)_{0.2}(\text{OH})_2$, $\text{Na}_3[\text{Co}(\text{CO}_3)_3]3\text{H}_2\text{O}$, $\text{Zn}_2(\text{CO}_3)$ (OH)₂, $\text{Bi}_2\text{Mg}(\text{CO}_3)_2(\text{OH})_4$, $\text{Fe}(\text{CO}_3)_{0.12}(\text{OH})_{2.76}$, $\text{Cu}_{1.54}\text{Zn}_{0.46}(\text{CO}_3)(\text{OH})_2$, $\text{CO}_{0.49}\text{Cu}_{0.51}$ (CO₃)_{0.43}(OH)_{1.1}, $\text{Ti}_3\text{Bi}_4(\text{CO}_3)_2(\text{OH})_2\text{O}_9(\text{H2O})_2$, and (BiO)₂CO₃.

(withdrawn) A gas generating composition as defined in claim, 103, wherein the co-oxidizer is a basic metal nitrate selected from the group consisting of Cu₂(OH)₃NO₃, CuCo(OH)₂NO₃, Zn₂(OH)₃NO₃, Mn(OH)₂NO₃, Fe₄(OH)₁₁NO₃·2H₂O₃.

(withdrawn) A gas generating composition as defined in claim 25, further comprising a carbon powder present from 0.1% to 6% by weight of the gas generating composition.

 $Mo(NO_3)_2O_2$, BiONO₂·H₂O, and Ce(OH)(NO₃)₃·3H₂O.

(previously presented) A gas generating composition as defined in claim 86, wherein the complex is selected from the group consisting of metal nitrate ammines.

wherein the release agent comprises graphite, molybdenum sulfide, calcium stearate or boron nitride.

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(amended) A solid gas generating composition formulated for generating gas suitable for use in deploying an air bag or balloon from a supplemental restraint system, the solid gas generating composition emprising consisting essentially of:

a complex of a metal cation and a neutral ligand containing hydrogen and nitrogen and sufficient oxidizing anion to balance the charge of the metal cation, wherein the complex is